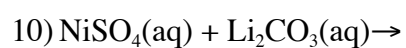
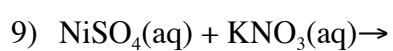
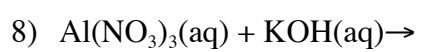
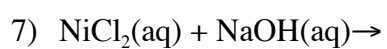
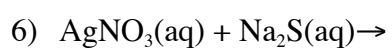
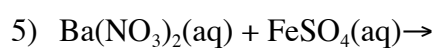
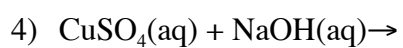
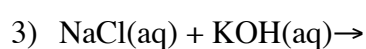
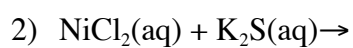
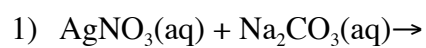


Name _____ Period _____

Precipitation Reactions

Write balanced molecular, complete ionic, and net ionic equations for the reactions.



11) Solutions of Nickel (II) sulfate and barium hydroxide are mixed

12) Solutions of Ammonium hydroxide and iron (III) chloride are mixed

13) Solutions of Magnesium sulfate and potassium nitrate are mixed

14) Solutions of Calcium chloride and sodium sulfate are mixed

15) Solutions of Ammonium sulfide and lead (II) nitrate are mixed

16) Solutions of Copper (II) chloride and sodium sulfate are mixed

17) Solutions of Barium hydroxide and zinc (II) sulfate are mixed

18) Solutions of Lead (II) nitrate and potassium iodide are mixed

19) Solutions of Iron (III) nitrate and sodium phosphate are mixed

20) Solutions of Barium hydroxide and copper (II) sulfate are mixed